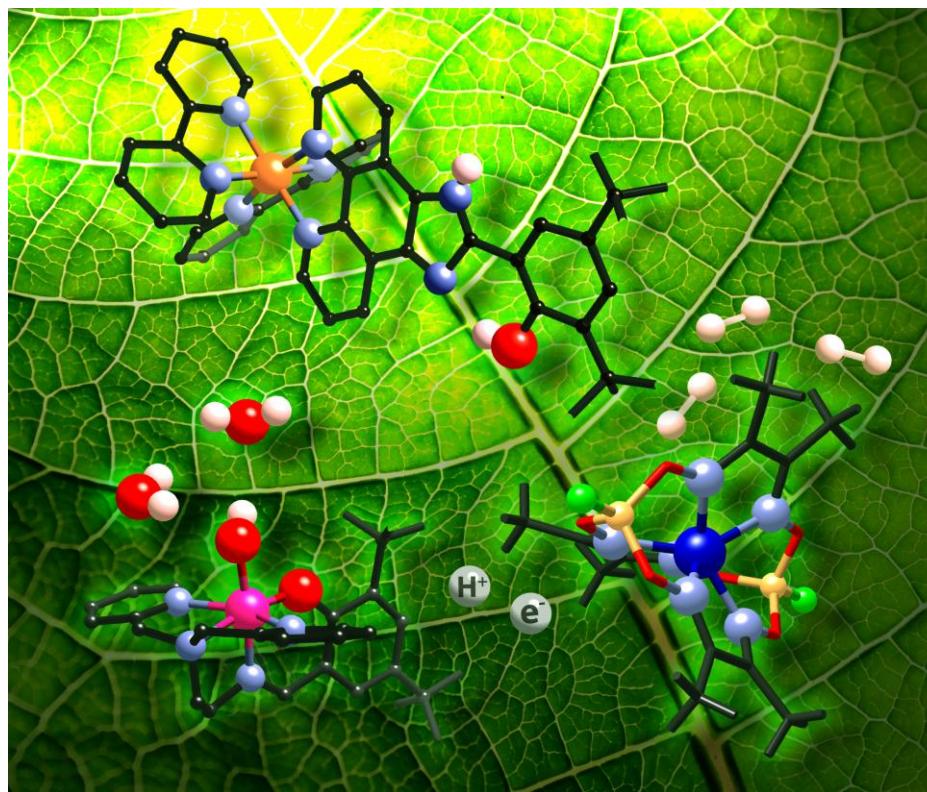


# Energie Bio-inspirée

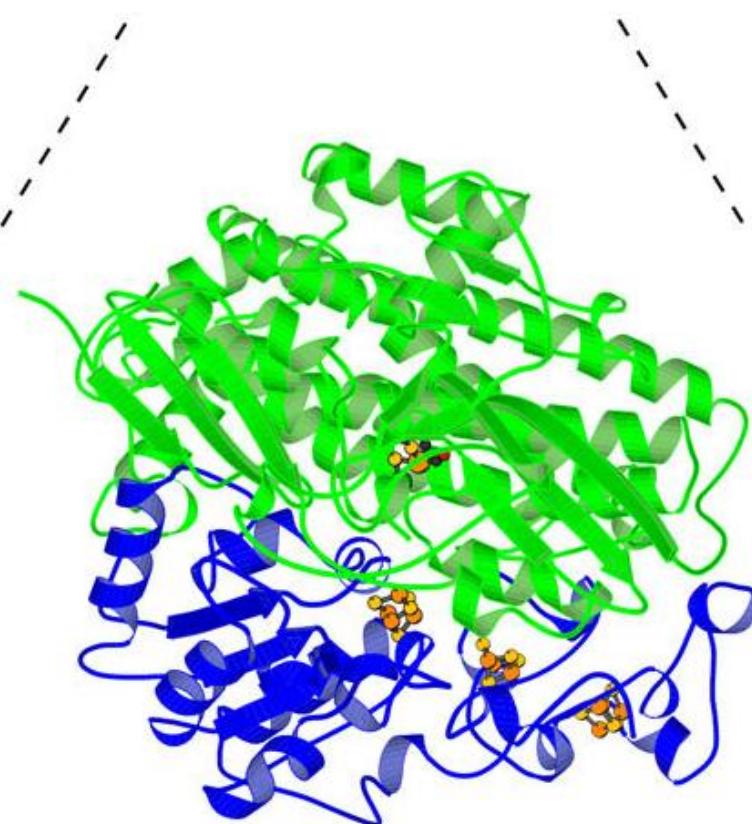
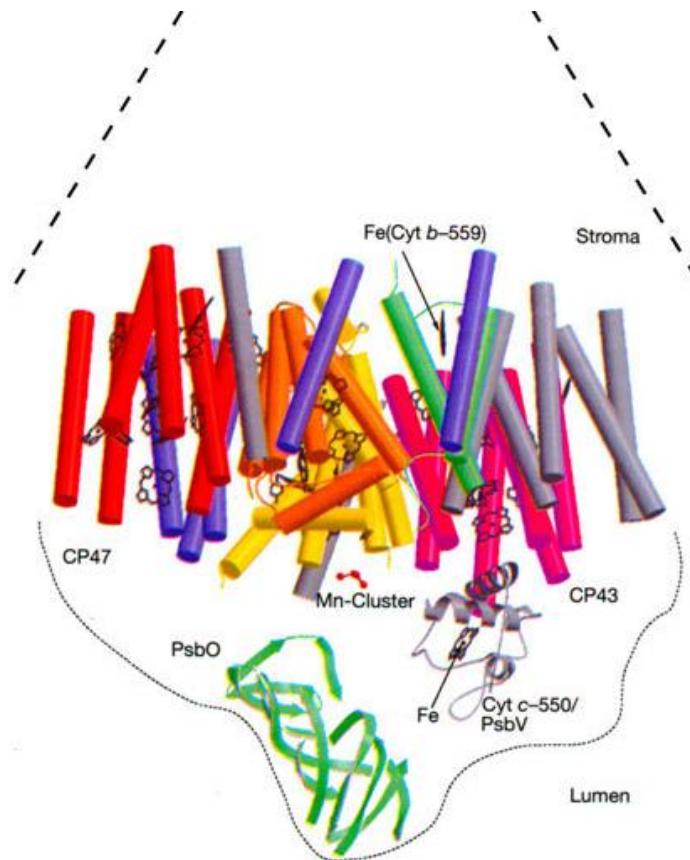
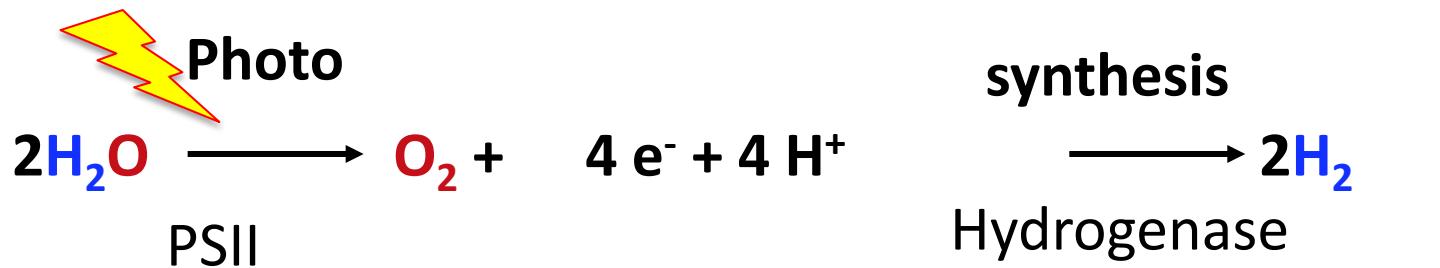


[ally.aukalo@u-psud.fr](mailto:ally.aukalo@u-psud.fr)

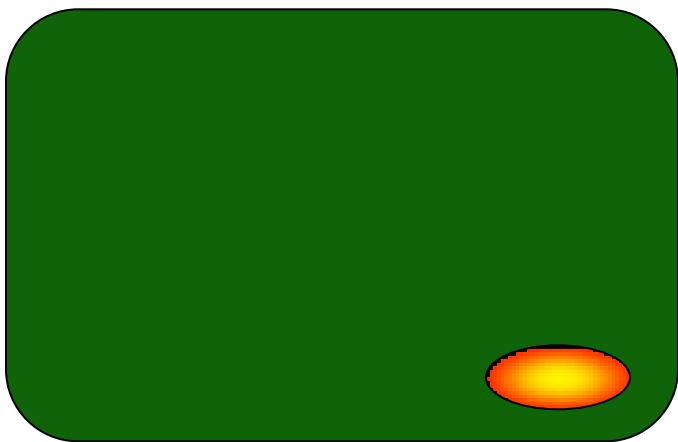
Université Paris Sud – Université Paris Saclay

Institut Joliot, CEA Saclay

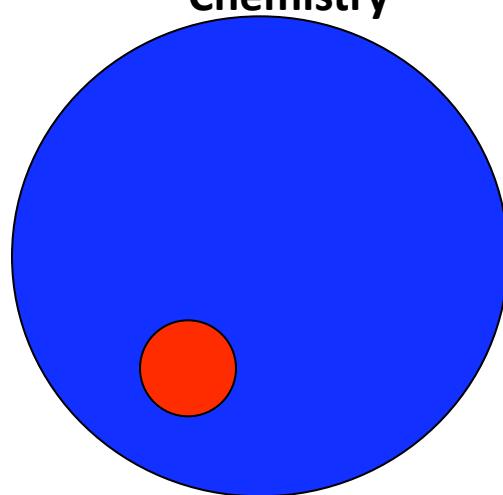
# Nature as source of inspiration



**Biology**

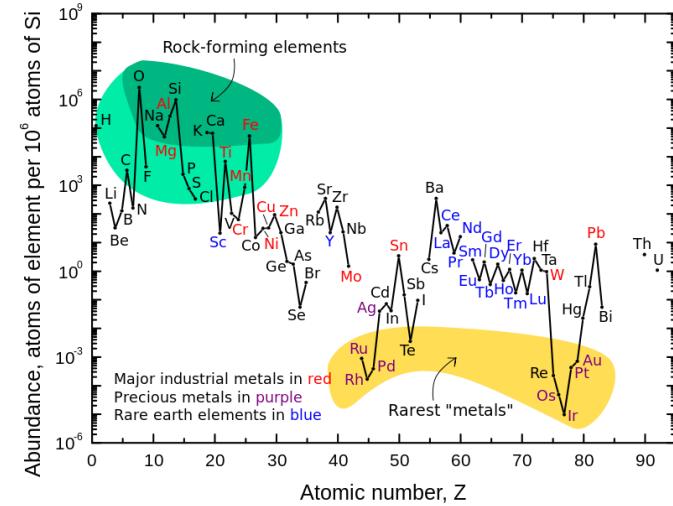
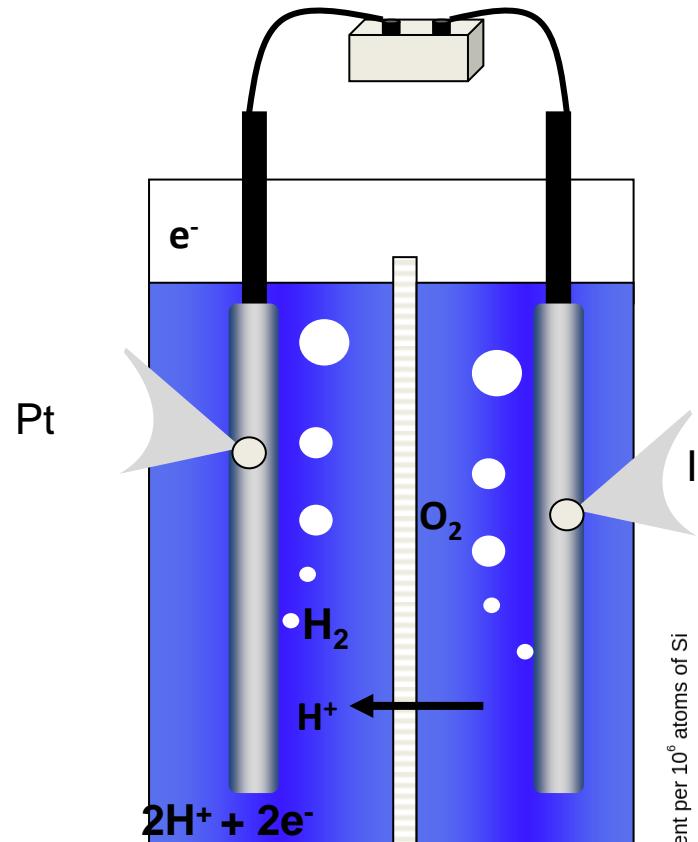


**Chemistry**

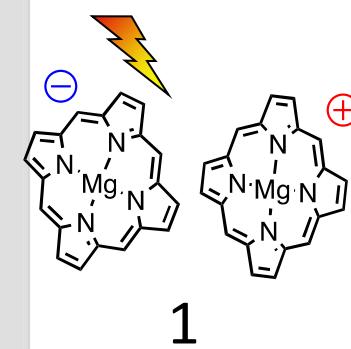
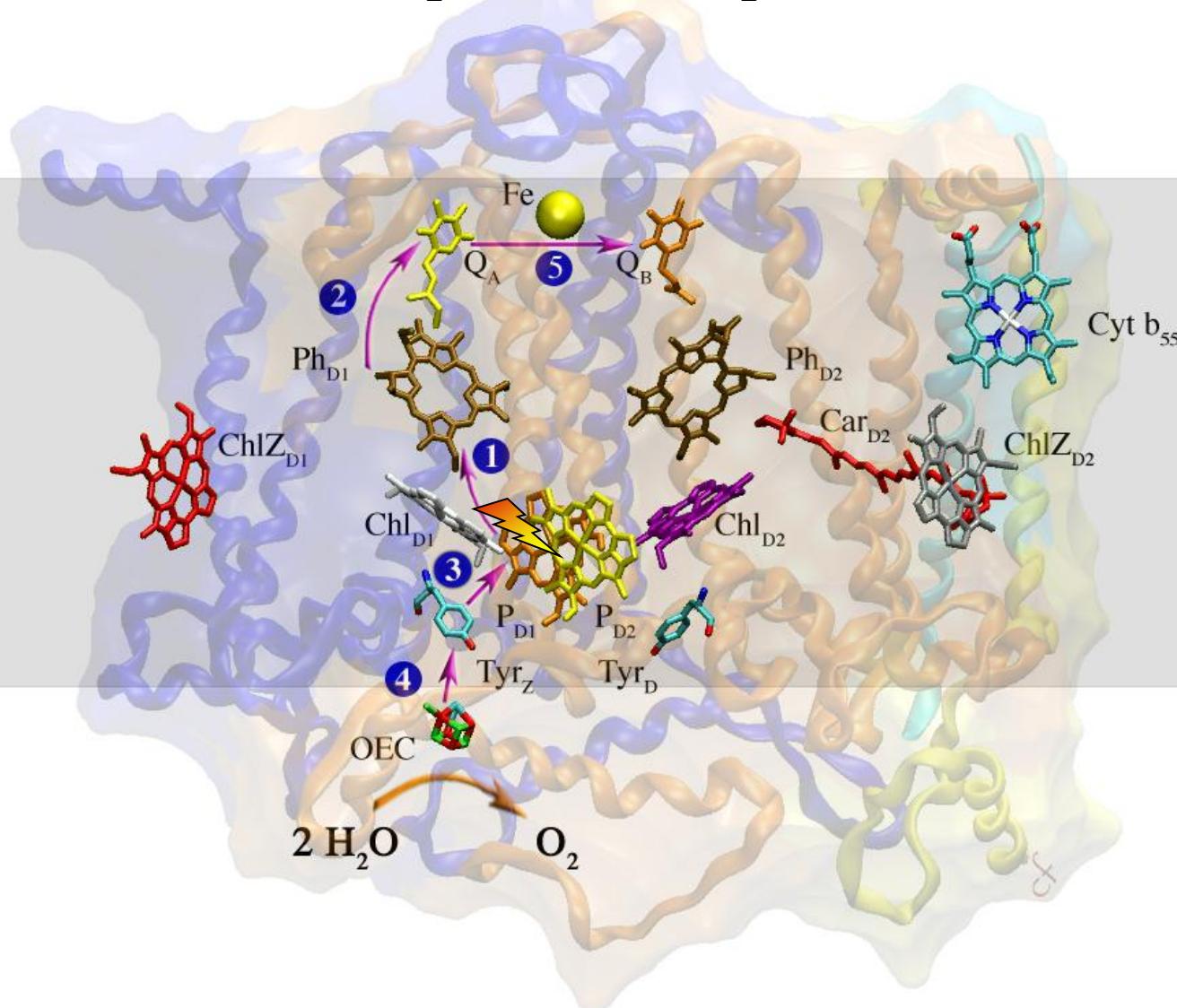


**Ridley's Paradox**

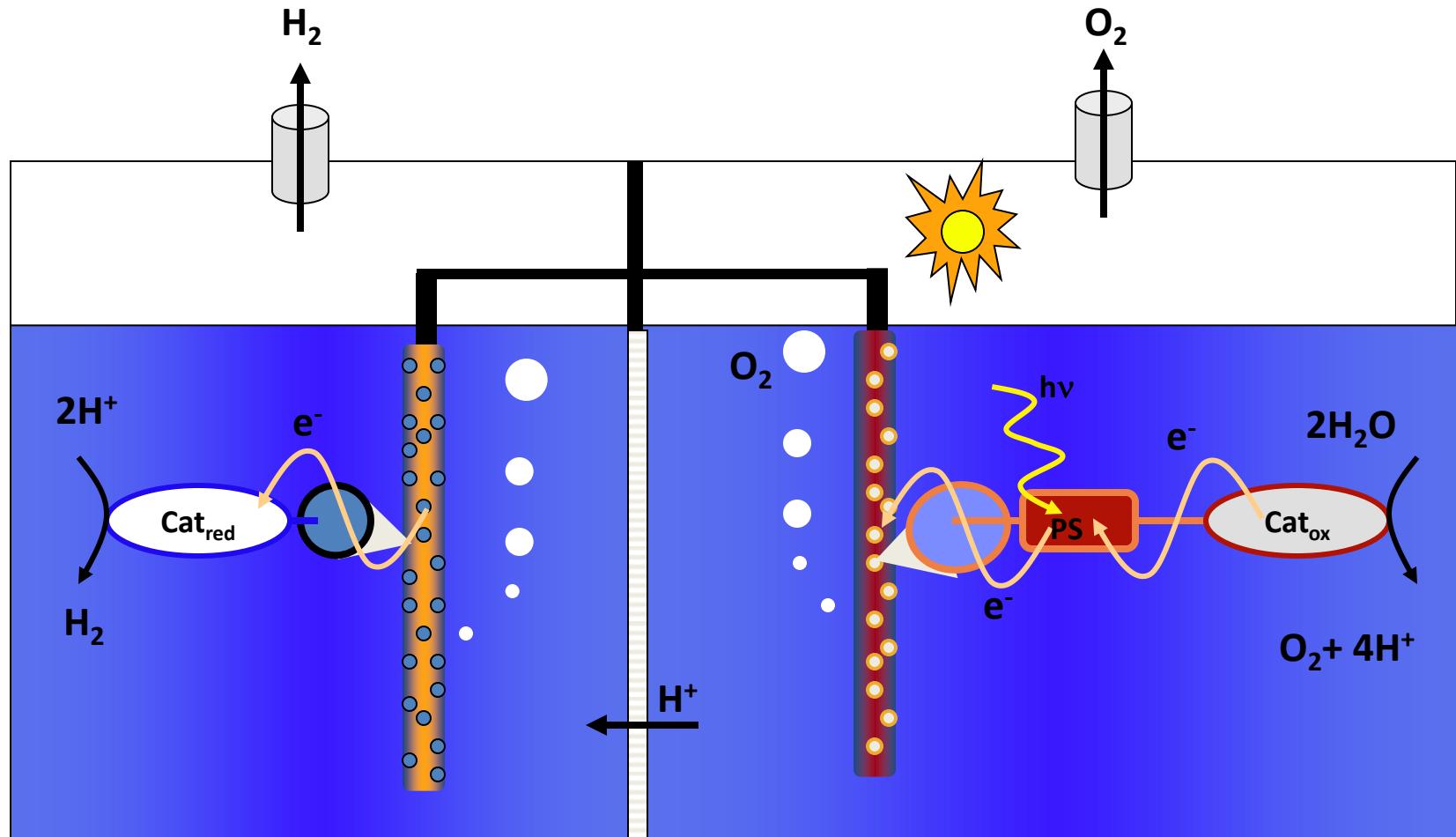
# Electrolyser



## Photosystem II



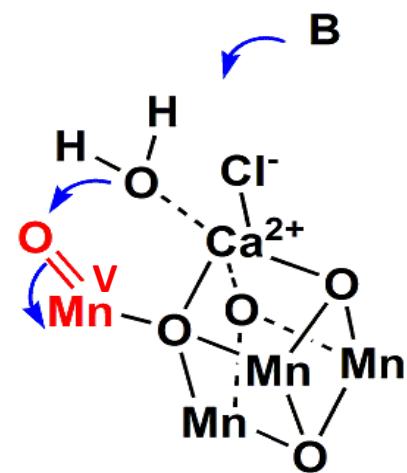
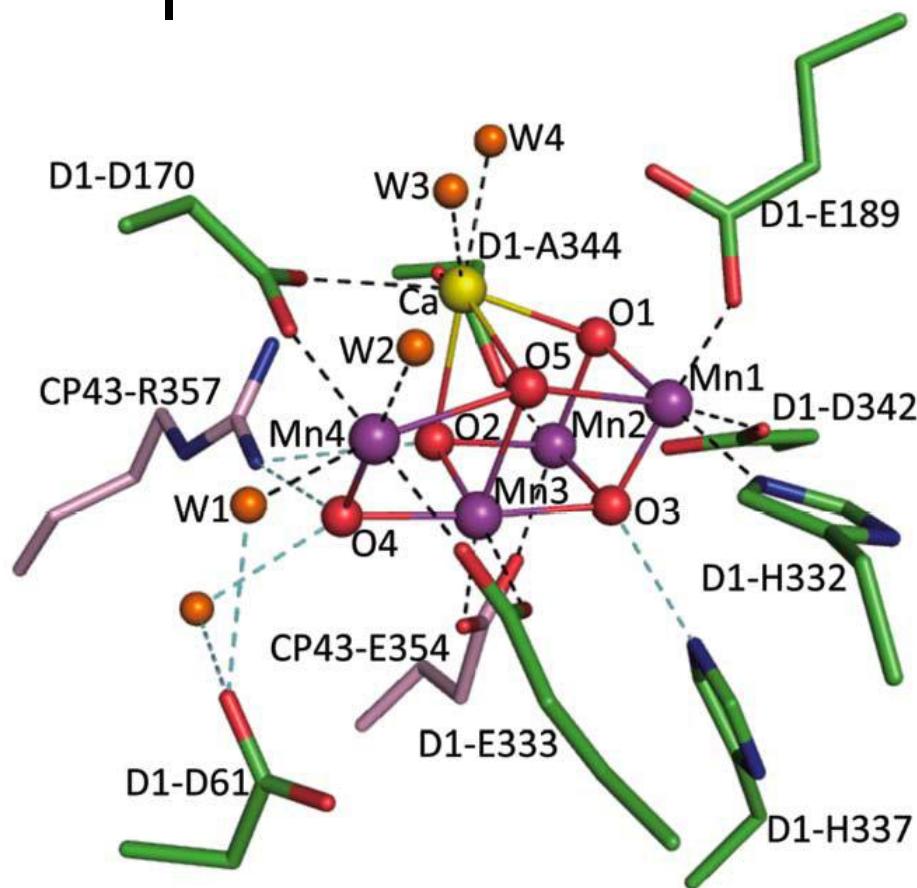
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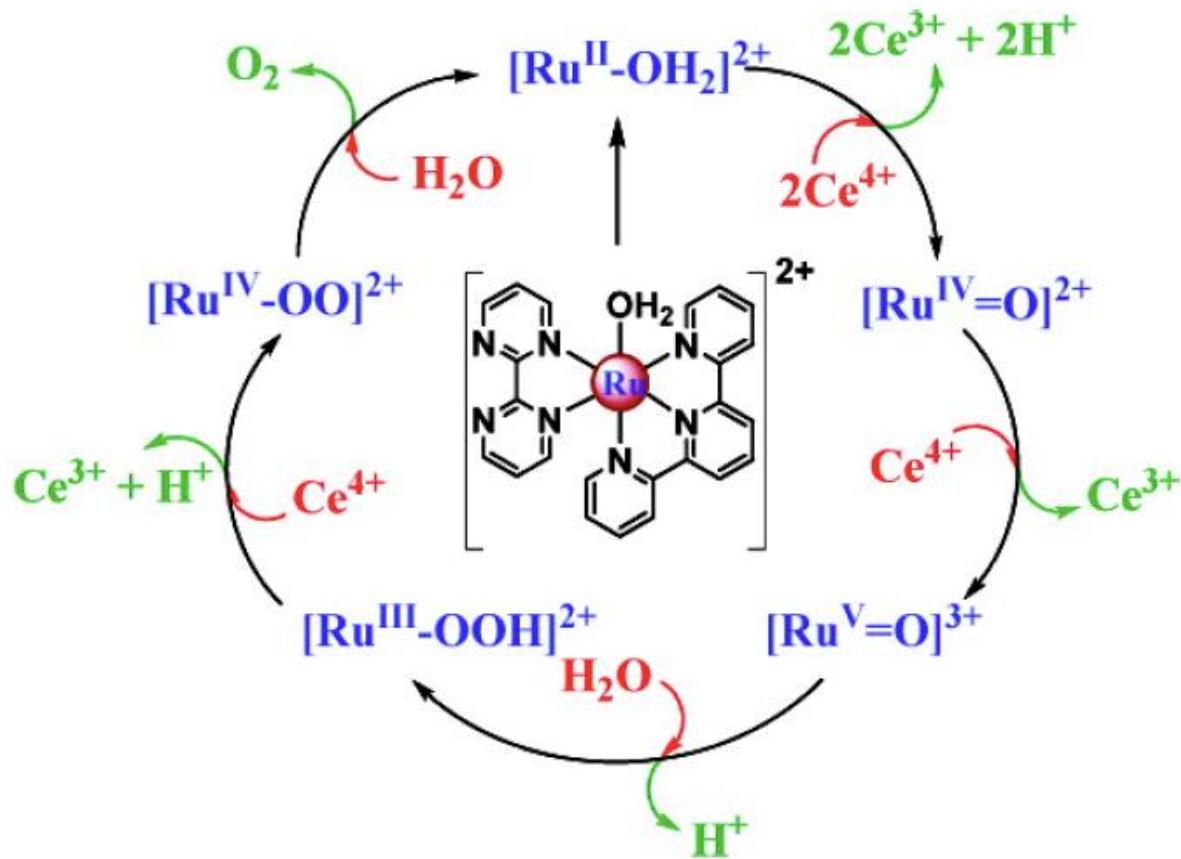
$Cat_{red}$ : Catalyst for reduction

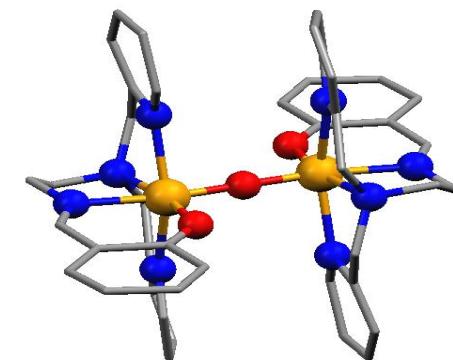
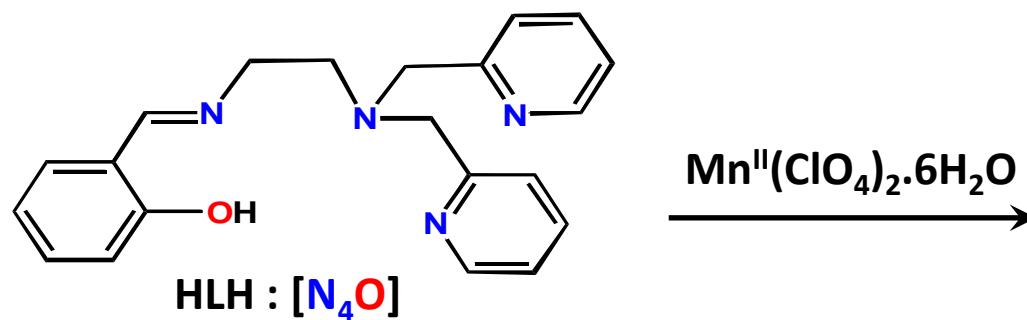
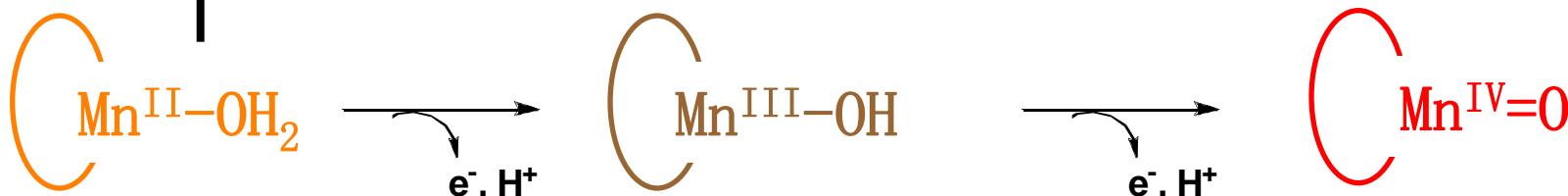
PS: Photosensitizer

$Cat_{ox}$ : Catalyseur for WO

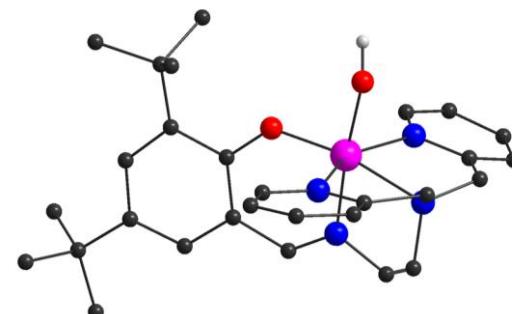
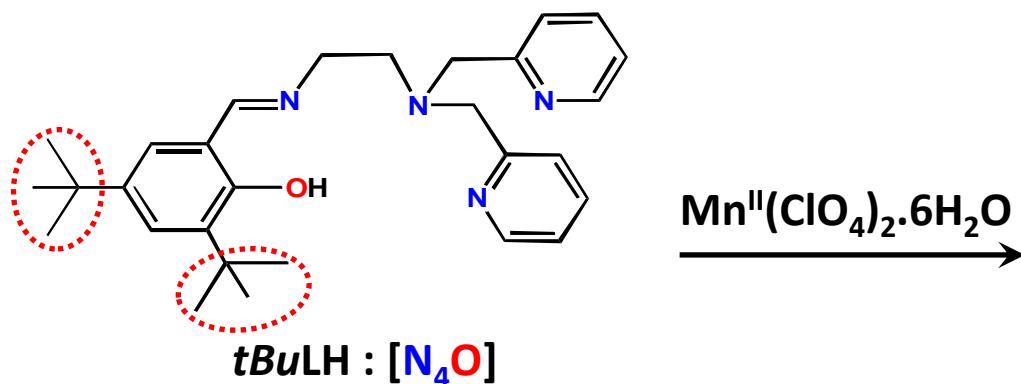


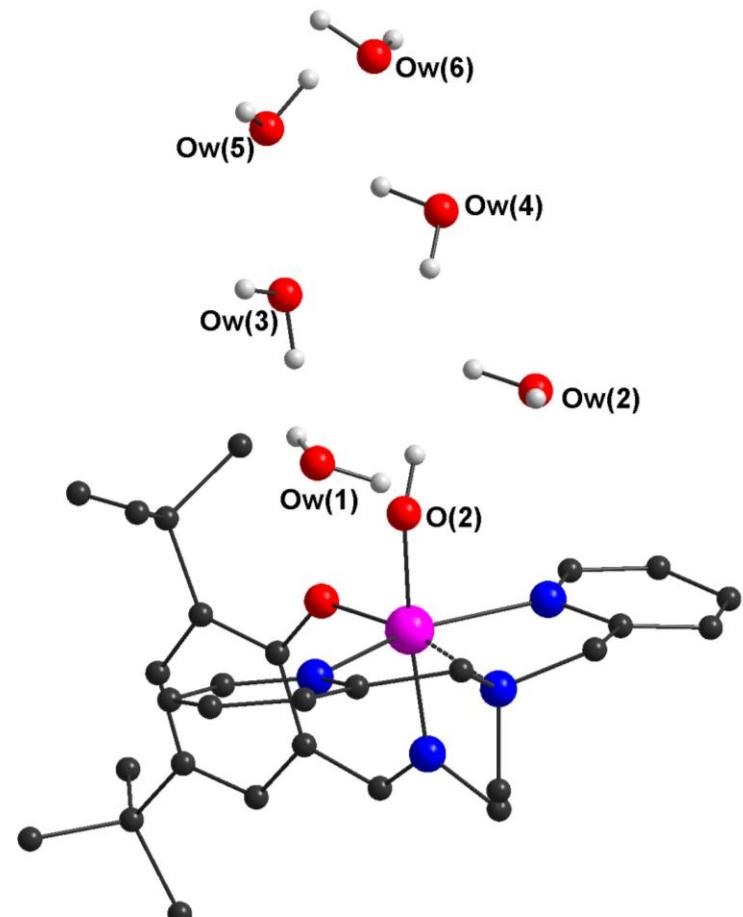
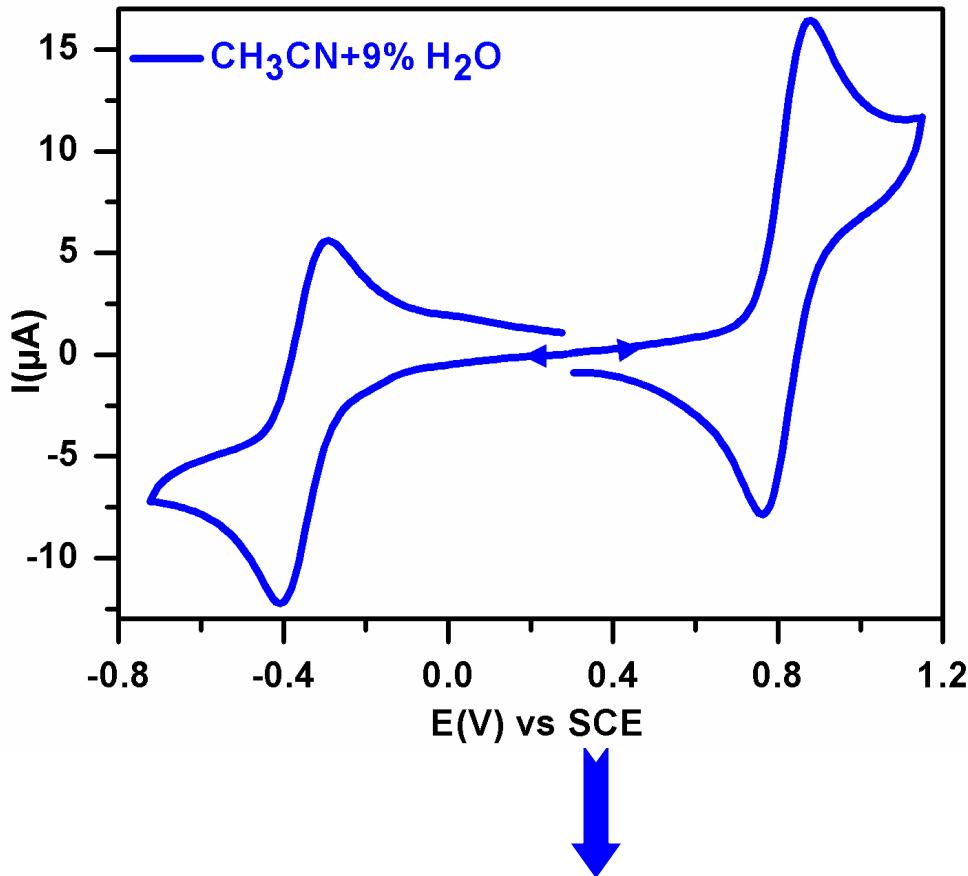
# Catalyseur pour l'oxydation de l'eau





O. Horner et al, Inorg.  
Chem. 1999



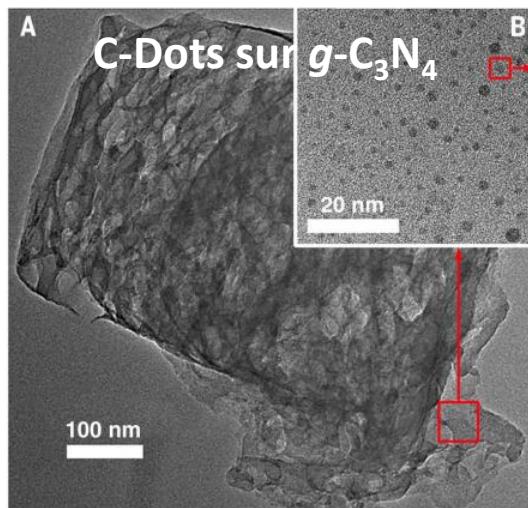
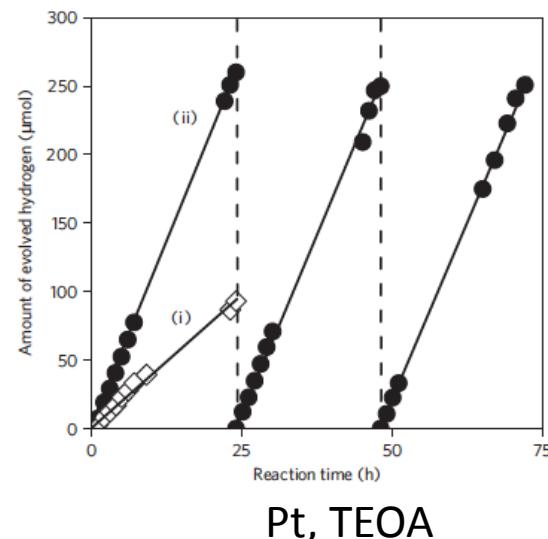
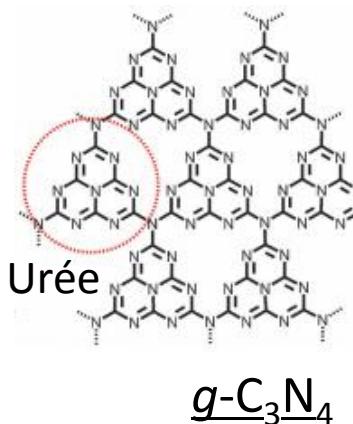


$\text{CH}_3\text{CN} / \text{H}_2\text{O}$

Gain in reversibility for both anodic and cathodic processes

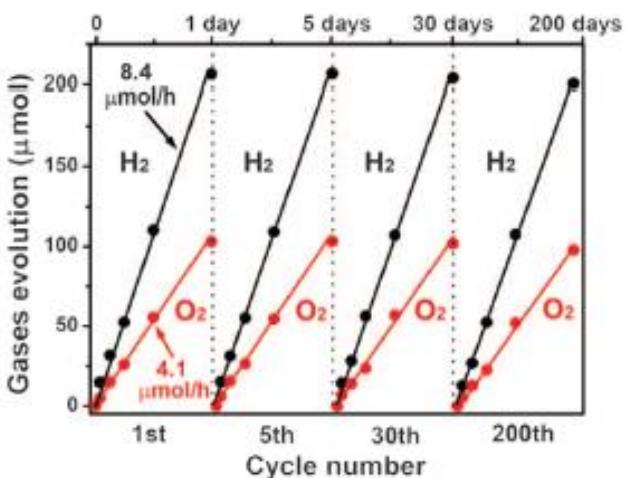
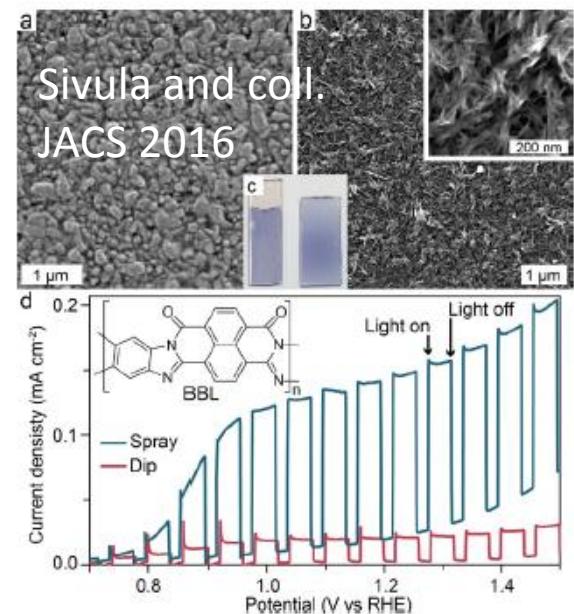
# New conjugated polymers for water splitting

M. Antonietti

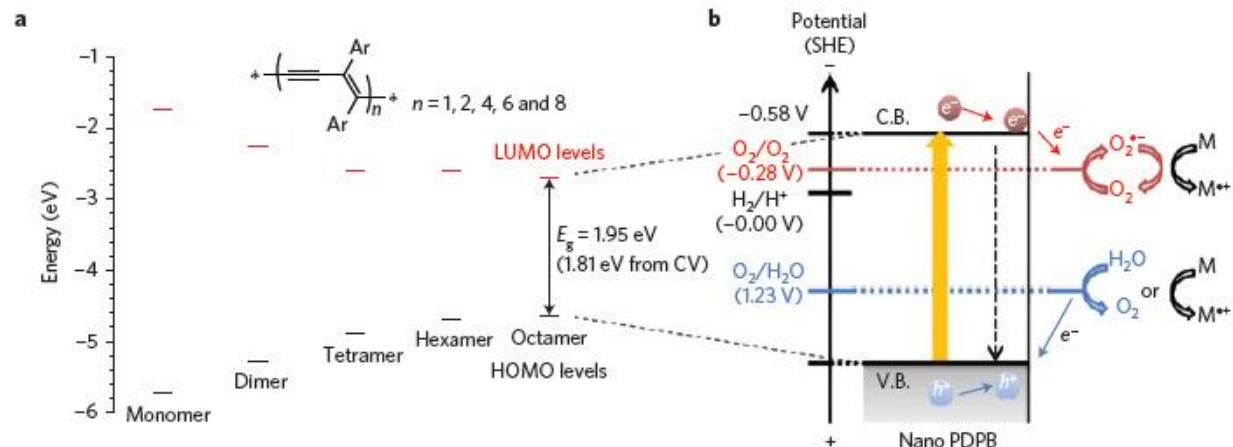
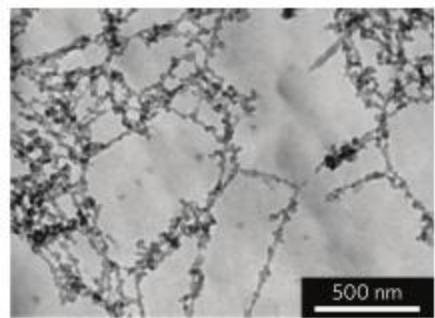


Water splitting  
under visible  
irradiation

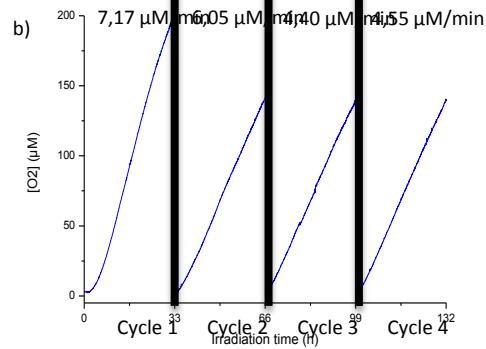
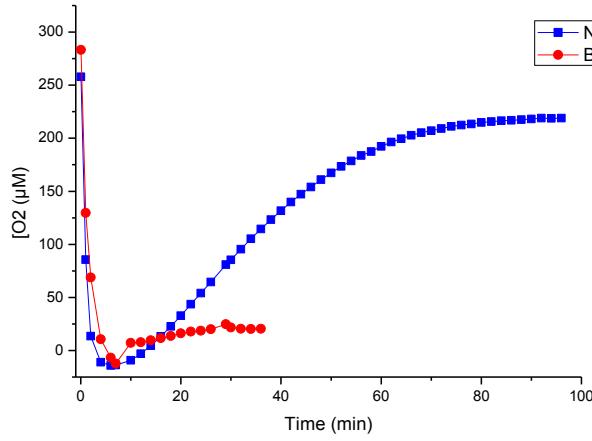
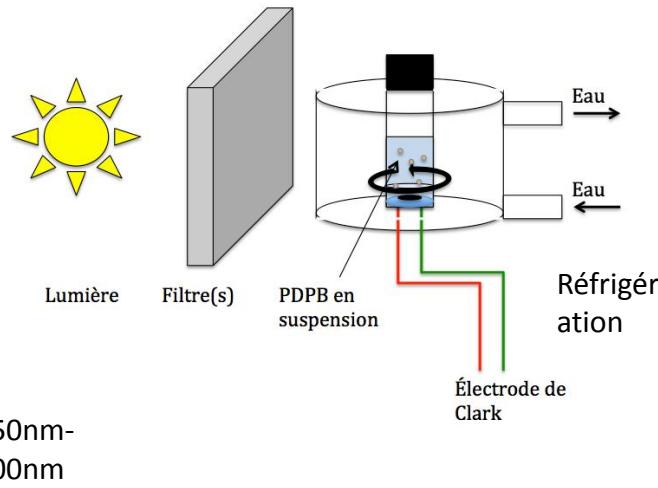
QE = 2% à 470 nm



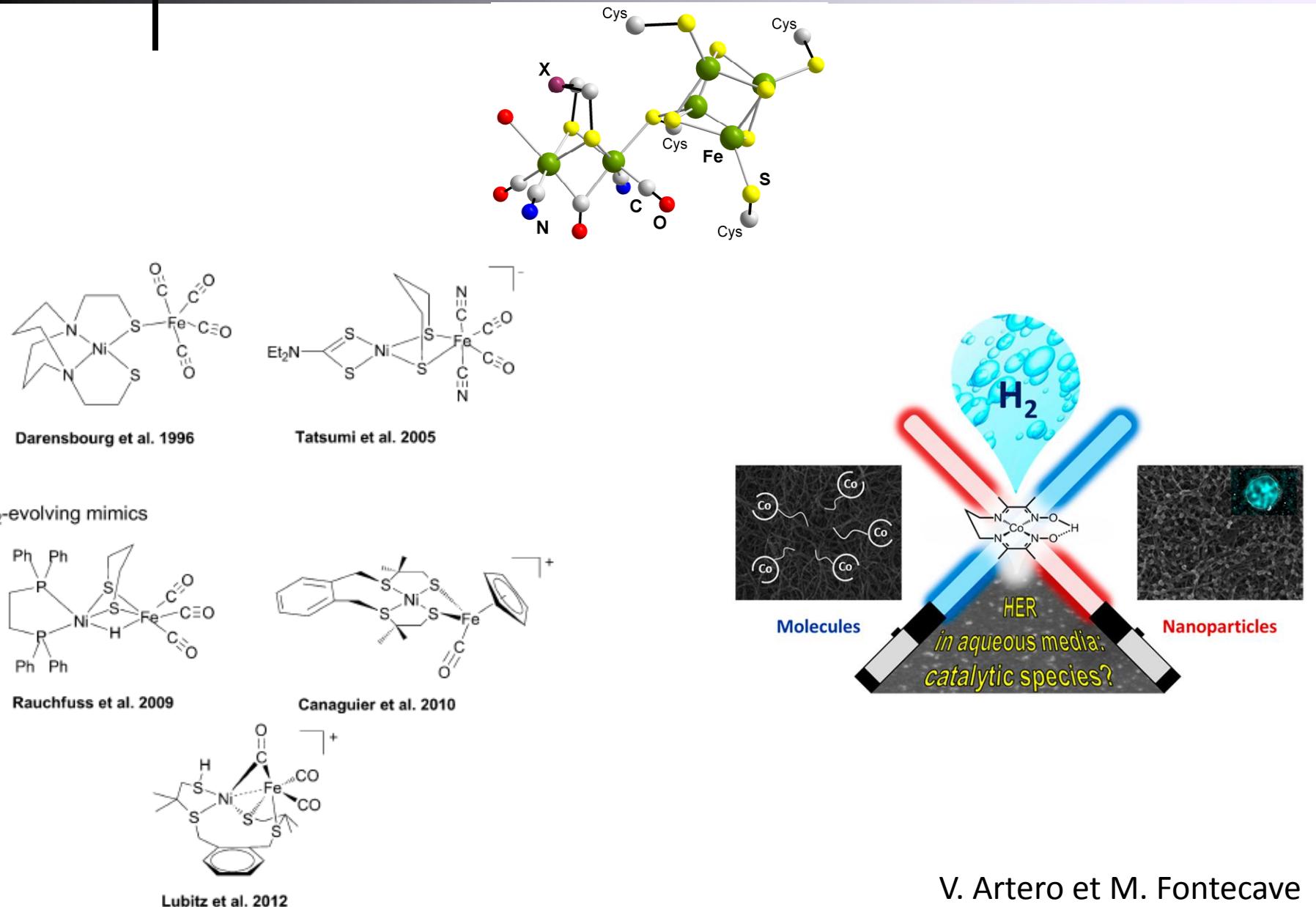
Liu and Kang, *Science*, 2015, 347, 970

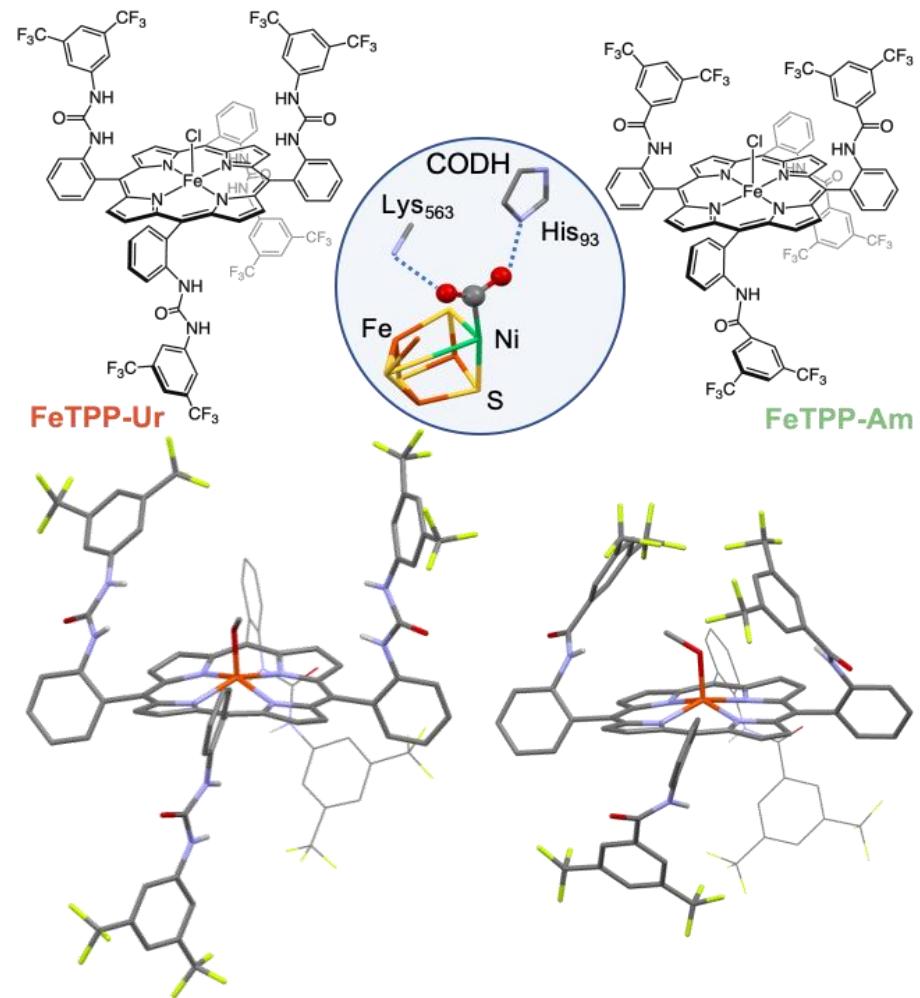
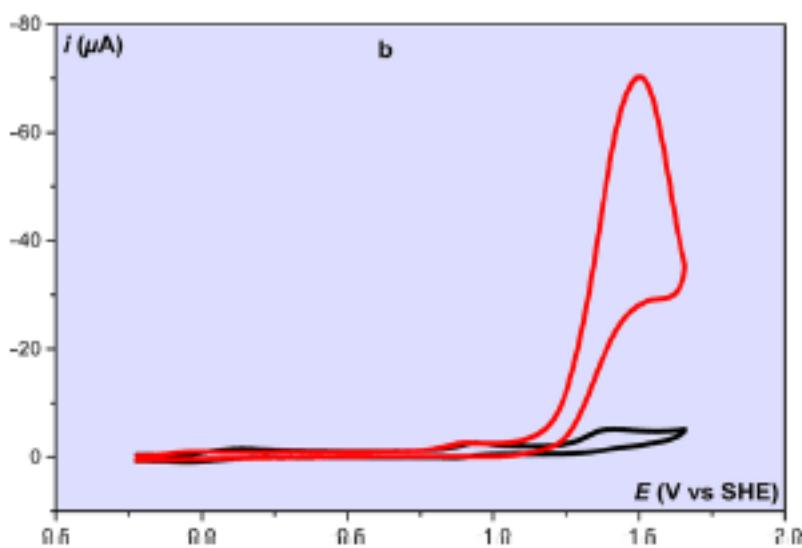
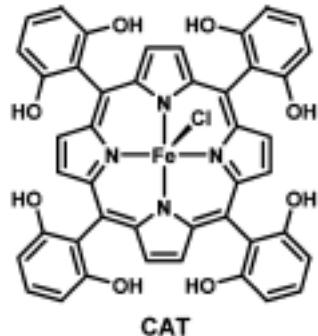


## Photocatalysis of water oxidation

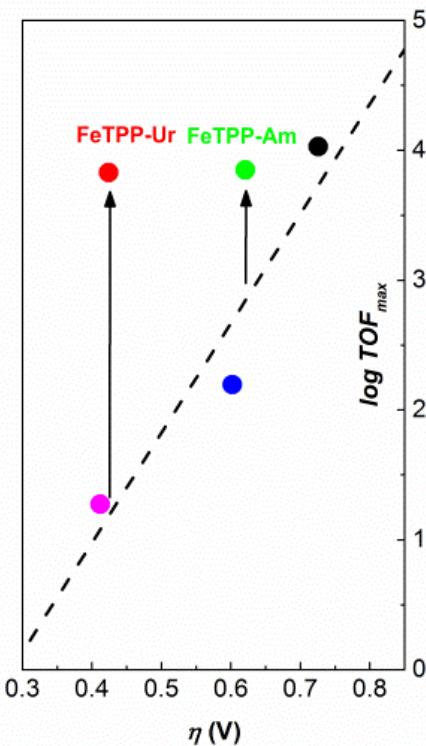
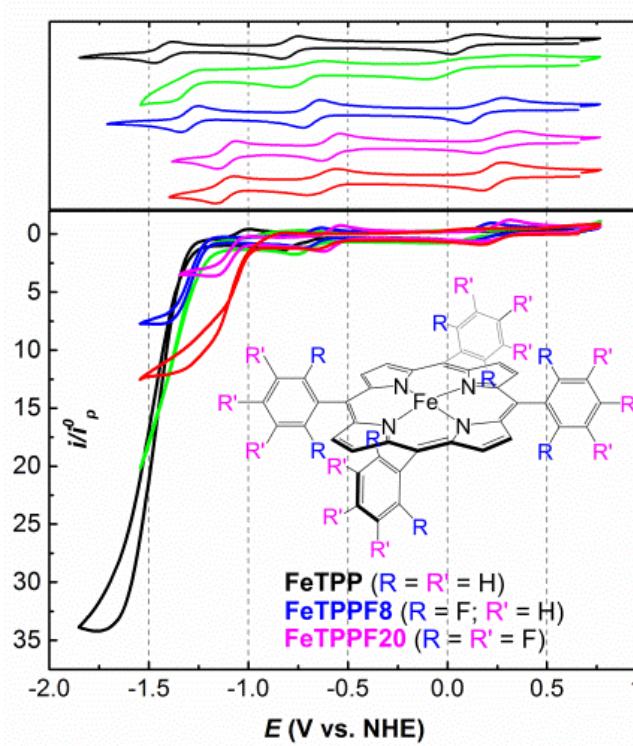
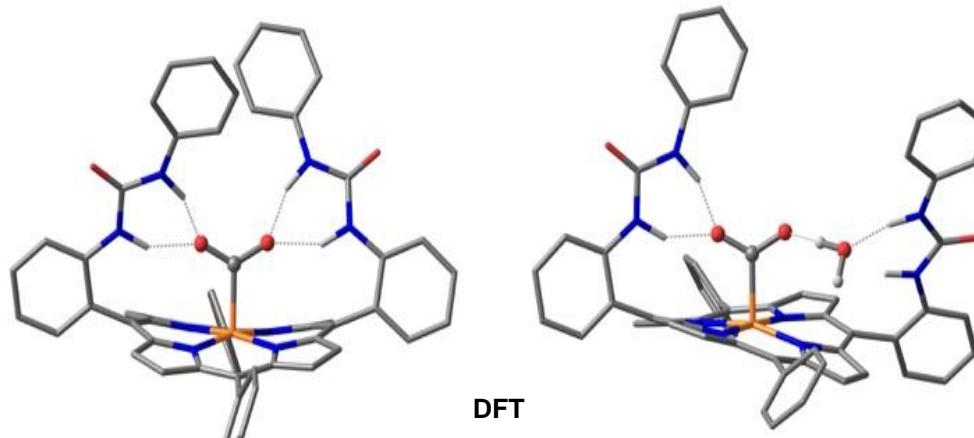


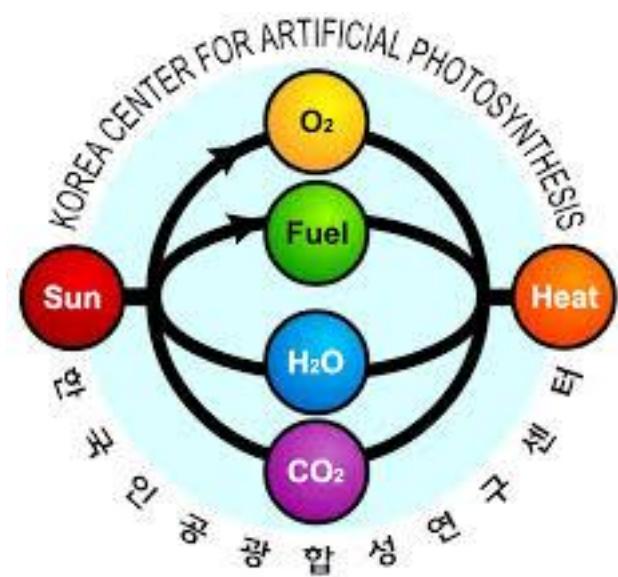
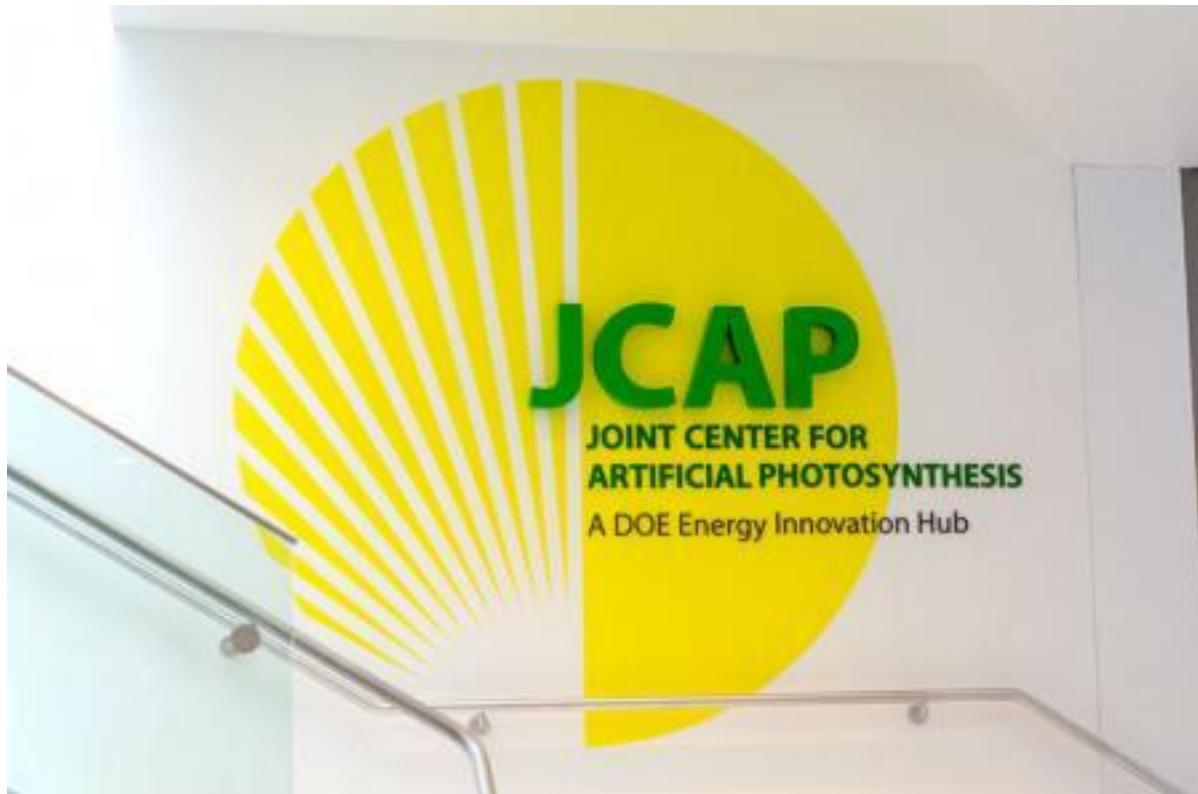
# Production de H<sub>2</sub>





# Réduction du CO<sub>2</sub>





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Tr.XAS

Microscopy

Cristallographic studies

Surface chemistry (PEM)

Theoretical study

Porphyrin Chemistry

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Thank you for your attention



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